

Synthesis of a Novel Series of 2-Substituted Imino-6-Amino-4-[2-Isobutoxy-5(4-Methyl-5-Carboxy-1,3-Thiazo-2-yl)]-Phenyl-1,3,5-Thiadiazines

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ABSTRACT

A novel series of 2-substituted imino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazines had been recently synthesized in this laboratory by refluxing 2-(3-substitutedthioamidoformamidino-4-isobutoxyphenyl)-4-methyl-5-carboxy-1,3-thiazoles with various isocyanodichlorides in acetone-ethanol medium in 1:1 molar proportion for 2 hours. The structures of all the synthesized compounds were justified on the basis of chemical characteristics, elemental analysis and spectral studies.

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Introduction

Heterocyclic compounds are more intriguing due to their utility in various fields. The literature survey reveals that when the compounds containing 1,3,5-dithiazino or 1,3,5-thiadiazino molecule as a parent nucleus then that molecule will enhance potency of that drug in medicinal, agricultural and industrial fields¹⁻⁹. Hence, nowadays the drug containing 1,3,5-dithiazino or 1,3,5-thiadiazino nucleus are widely used in the drug¹⁰⁻¹⁶. It has been reported that thiadizino nucleus and its derivatives possesses antiviral, antifungal, antibacterial, anti-tuberculostatic and anti-helminthic activities¹⁷⁻¹⁸. Several thiadiazines are used in the treatment of cancer¹⁹ and some are anti-HIV²⁰⁻²¹ drugs. They are also used in agriculture²² as fungicidal²³, insecticide²⁴ while 1,3,5-dithiazines are also effective against copper corrosion²⁵ and used in lubricating oil²⁶. The various researchers have been investigated briefly essential reactions of substituted isocyanodichlorides²⁷⁻³². In this laboratory various alternative routes for the synthesis of nitrogen, nitrogen and sulphur containing heteroacycles and heterocycle had been developed. It is quite intriguing to investigate one step cyclisation of 2-(3-substitutedthioamidoformamidino-4-isobutoxyphenyl)-4-methyl-5-carboxy-1,3-thiazoles (IIIa-e) with N-substitutedisocyanodi-chlorides (VIIa-h) in acetone-ethanol medium to isolate 2-substituted imino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazines (VIIa 1-e 37).

Materials and Methods

The melting points of all the synthesized compounds were recorded using hot paraffin bath and are uncorrected. The purity of the compounds was checked on Silica Gel-G plates by TLC with layer thickness of 0.3 mm. All chemicals used were of AR grade (Indiamake). Alkyl/aryl isothiocyanates, isothiocarbamoylchloride, isocyanodichlorides and phenylthiourea have been prepared by known literature method. The carbon and hydrogen analysis was carried out on Carlo-Ebra-1106 analyser. Nitrogen estimations were carried out on Colman-N-analyser-29.

IR spectra were recorded on Perkin-Elmer spectrometer in the range 4000-400 cm⁻¹ in KBr pellets.

PMR spectra were recorded on Bruker AC-400F spectrometer with TMS as internal standard using CDCl₃ and DMSO-d₆ as solvent.

MASS spectra were recorded on WATERS, Q-TOFmicro mass (LC-MS).

During the XRD analysis of the compounds, configuration is equal to reflection spinner stage having owner equal to jagtar. Goniometer is equal to PW 3050/60 Theta minimum step size is 0.001 Omega. The diffractometer system which is used during the analysis is of XPERT-PRO. Scan axis is Gonio. Carbon hydrogen and nitrogen analysis and all spectral analysis were carried out at P. U.Chandigarh.

Result and Discussion

Synthesis of 2-phenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIa4)

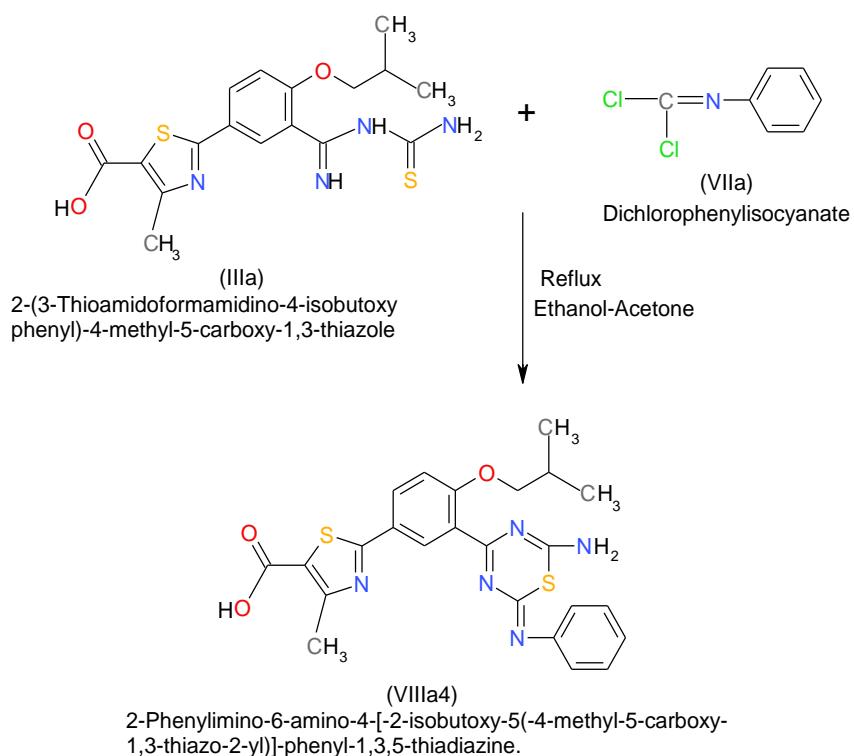
The interaction of 2-(3-thioamido-formamidino-4-isobutoxyphenyl)-4-methyl-5-carboxy-1,3-thiazole (IIIa) with phenylisocyanodichloride(VIIa) in 1:1 molar ratio was refluxed on water bath in acetone-ethanol medium for 2 hours. During heating evolution of hydrochloride gas was clearly noticed. After distillation of excess of acetone-ethanol ivory colour product was isolated which on basification with dilute ammonium hydroxide white crystalline products afforded, yield 88%, m.p. 158°C.

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The probable mechanism for the formation of (VIIIa4) is depicted below,

Reaction



Properties of (VIIIa4)

1. It is white crystalline solid having m.p. 158°C.
2. It gave positive test for nitrogen and sulphur.
3. It gave positive test for carboxylic group.
4. It does not desulphurized when boiled with sodiumplumbite solution which is clearly indicates that sulphur is not free and gets cyclised³³⁻³⁴.
4. It was soluble in benzene, acetic acid, DMF and DMSO.

5. Elemental analysis

The result of elemental analysis is given in Table No. V-1

Table No. V-1

Elements	Found (%)	Calculated (%)
Carbon	57.50	58.41
Hydrogen	04.00	04.66
Nitrogen	14.91	14.91
Sulphur	11.62	12.98

6. From the analytical data the molecular formula was found to be C₂₅H₂₅N₅O₃S₂.

7. IR Spectrum

The IR spectrum of compound was carried out in KBr pellets and reproduced on IRPlate No. PPB-14 an important absorption are correlated as follows in Table No.V-2.

Table No. V-2

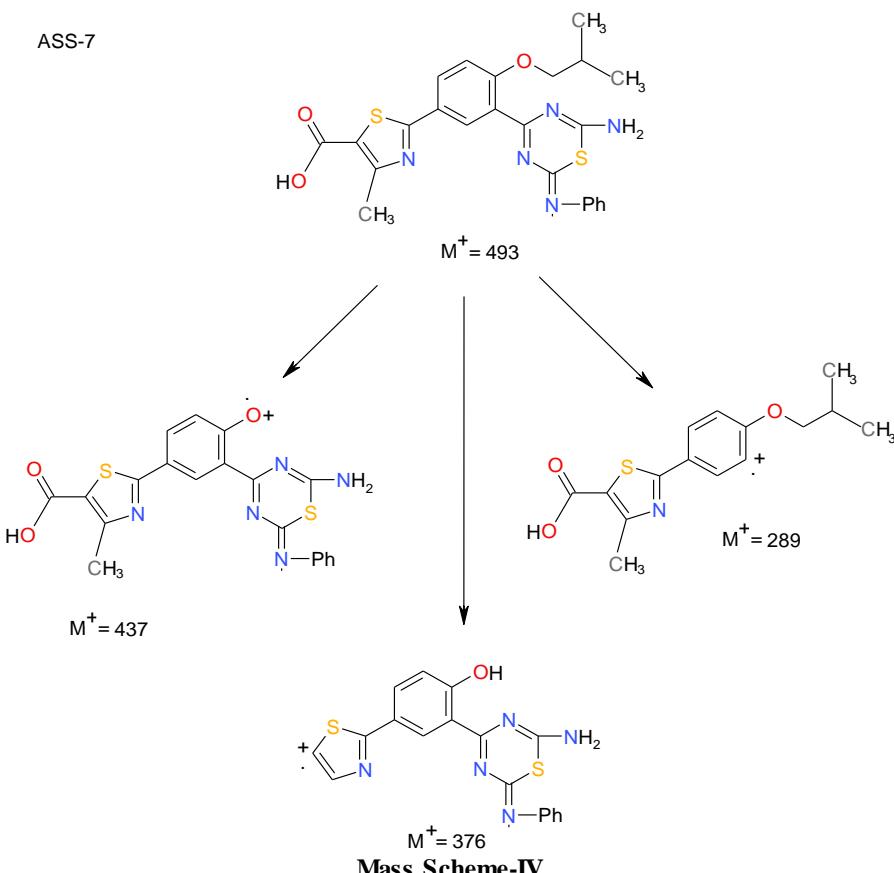
Absorption Observed (cm ⁻¹)	Assignment ³⁵⁻³⁷	Absorption Expected (cm ⁻¹)
3460.48	-NH Stretching	3500-3000
3105.48	-C-Hstretching	3150-3000
2228.58	-S-C=Nstretching	2270-1940
1706.17	-C=Nstretching	1750-1450
1285.23	-C-N stretching	1360-1000
0738.42	-C-Sstretching	800-600

PMR-Spectrum

1. The PMR spectrum³⁸ of compound was carried out in CDCl₃ and DMSO-d₆ and reproduced on PMR Plate No. PPB-14 This spectrum distinctly displayed the signals due -COOH proton at 9.9481 ppm, Ar-H protons at 8.2077-7.9343 ppm, monosubstituted phenyl protons at 7.1516-7.1293 ppm, -NH₂ protons at δ 3.6368-3.4143 ppm, -OCH₂ protons at δ 2.5457-2.4074 ppm, -CH proton at δ 1.9079-1.8622 ppm, -CH₃ protons at δ 0.9935-0.6244 ppm.

2. Mass spectrum

The Mass analysis of compound VIIIa4 was carried out and reproduced on Mass Plate No. ASS-7. The fragmentations occur during the analysis is given in Mass Scheme-IV.



3. XRD Analysis

The XRD Analysis of compound **VIIa4** was carried out, during analysis the start position is 02 Th which shows reading from 5.0083 and end position 02 Th is 69.9984. It takes 29.845 second. The analysis of this compound was carried out at 25°C. Copper is used as anode material. The result and values obtained during XRD are as given below,

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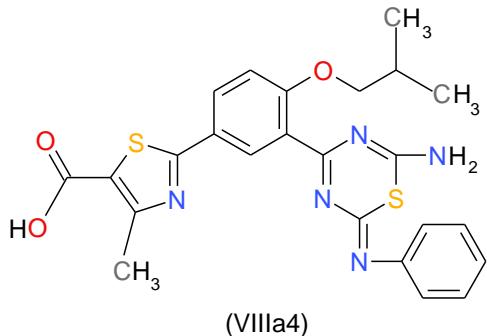
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From the above properties and spectral analysis of the compound (VIIIa4) was assigned the structure as 2-phenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIIa4)

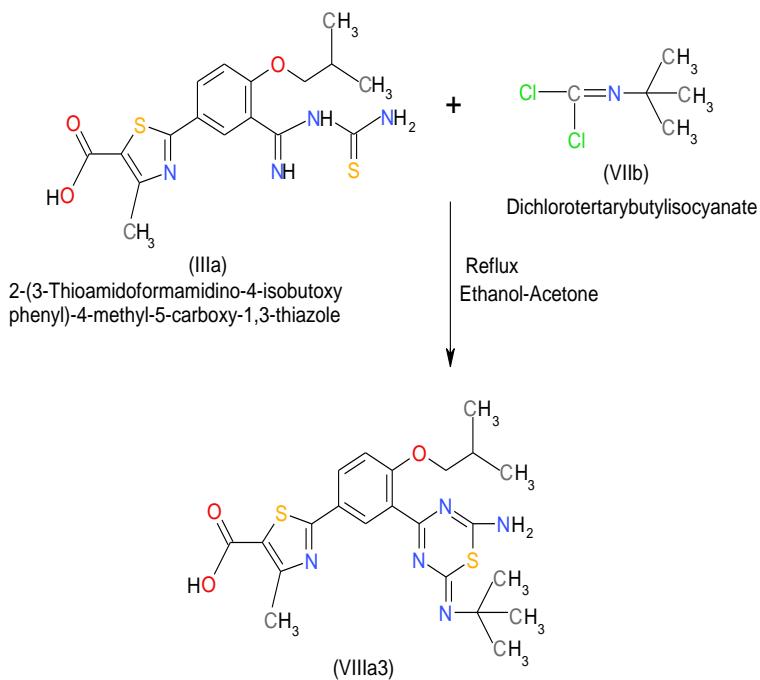


2-Phenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine.

Synthesis of 2-tert-butylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIIa3)

The interaction of 2-(3-thioamidoformamidino-4-isobutoxyphenyl)-4-methoxy-5-carboxy-1,3-thiazole (IIIa) with tert-butylisocyanodichloride(VIIb) in 1:1 molar ratio was refluxed on water bath in acetone-ethanol medium for 2 hours. During heating evolution of hydrochloride gas was clearly noticed. After distillation of excess of acetone-ethanol red coloured product was isolated which on basification with dilute ammonium hydroxide brown crystals were afforded, yield 84%, m.p.166°C.

Reaction



Properties of (VIIIa3)

1. It is white crystalline solid having m.p.166°C.
2. It gave positive test for nitrogen and sulphur.
3. It gave positive test for carboxylic group.
4. It does not desulphurized when boiled with sodium plumbite solution which clearly indicates that sulphur is not free and gets cyclised^{157,230}.
5. It was soluble in benzene, acetic acid, DMF and DMSO.

6. Elemental analysis

The result of elemental analysis is given in Table No. V-3

Table No. V-3

Elements	Found (%)	Calculated (%)
Carbon	54.65	55.81
Hydrogen	04.90	05.70
Nitrogen	14.79	14.79
Sulphur	12.10	13.53

7. From the analytical data the molecular formula was found to be $C_{23}H_{29}N_5O_3S_2$.

8. IR Spectrum

The IR spectrum of compound was carried out in KBr-pellets and reproduced on IRPlate No. PPB-15, an important absorption are correlated as follows in Table No. V-4

Table No. V-4

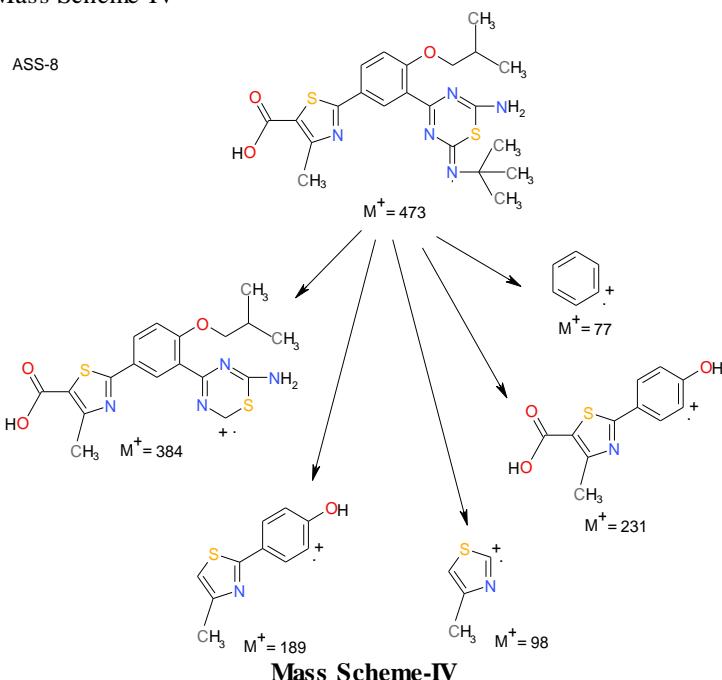
Absorption observed (cm^{-1})	Assignment ^{231,259-260}	Absorption Expected (cm^{-1})
3452.25	-NH Stretching	3500-3000
3000.00	(Ar)C-Hstretching	3150-3000
1646.47	-C=Nstretching	1750-1450
1634.00	(Ar)C=Cstretching	1600-1450
1107.17	-C-N stretching	1360-1000
0701.48	-C-Sstretching	0800-0600

9. PMR-Spectrum

The PMR spectrum²⁶¹ of compound was carried out in CDCl_3 and DMSO-d_6 and reproduced on PMR Plate No. PPB-15. This spectrum distinctly displayed the signals due to -COOH proton at δ 8.32 ppm, Ar-H protons at δ 6.5049-6.4889 ppm, -NH₂ protons at δ 4.9238-4.0207 ppm, -CH₃ protons at δ 1.1894-1.2322 ppm, -O-CH₂ protons at δ 3.0088-3.8509 ppm, -CH proton at δ 0.8599-0.8432 ppm.

10. Mass spectrum

The Mass analysis of compound VIIa3 was carried out and reproduced on Mass Plate No. ASS-8. The fragmentations occur during the analysis is given in Mass Scheme-IV



11. XRD Analysis

The XRD Analysis of compound VIIa3 was carried out, during analysis the start position is 02 Th which shows reading from 5.0083 and end position 02 Th is 69.9984. It takes 29.845 second. The analysis of this compound was carried out at 25°C. Copper is used as anode material. The result and values obtained during XRD are as given below,

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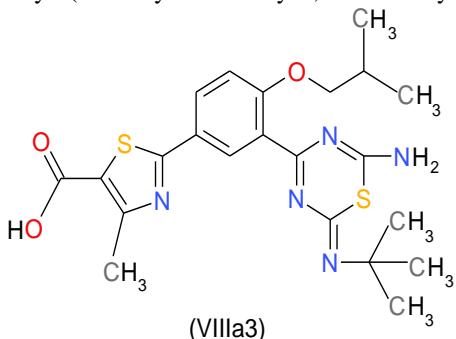
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From the above chemical characteristics, elemental and spectral analysis the compound (VIIIa3) was assigned the structure as 2-tert-butylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIIa3)



2-Tert-butylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine.

Similarly, 2-(3-thioamidoformamido)-4-isobutoxyphenyl-4-methyl-5-carboxy-1,3-thiazole (IIIa), were interacted with methylisocyanodichloride(VIIa),ethylisocyanodichloride(VIIb)p-cl-phenylisocyanodichloride (VIIe)o-tolylisocyanodichloride (VIIf),m-tolylisocyanodichloride(VIIg),p-tolylisocyanodichloride(VIIh),by above mentioned method to isolate 2-methylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIIa1), 2-ethylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine (VIIIa2), 2-p-chlorophenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine (VIIIa5), 2-o-tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine (VIIIa6), 2-m-tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine (VIIIa7) and 2-p-tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine(VIIIa8)respectively and enlisted in Table No. V-5.

Table No. V-5

	Compd. No.	2-Substitutedimino 6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	Yield %	M.P. °C
1. (VIIIa1)	2-Methylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	85	150	
2. (VIIIa2)	2-Ethylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	83	155	
3. (VIIIa3)	2-Tert-butylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	84	166	
4. (VIIIa4)	2-Phenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	88	158	
5. (VIIIa5)	2-p-Chlorophenylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	85	165	
6. (VIIIa6)	2-o-Tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	82	160	
7. (VIIIa7)	2-m-Tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	82	162	
8. (VIIIa8)	2-p-Tolylimino-6-amino-4-[2-isobutoxy-5(4-methyl-5-carboxy-1,3-thiazo-2-yl)]-phenyl-1,3,5-thiadiazine	80	176	

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